

DRAFT WORKING DOCUMENT FOR COMMENTS:

WHO good manufacturing practices considerations for the prevention and control of nitrosamine contamination in pharmaceutical products

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39 40 WHO good manufacturing practices considerations for the prevention and control of nitrosamine contamination in pharmaceutical products.

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Review and finalization of the first draft working document with an informal drafting group.	February 2024
Mailing of working document to the Expert Advisory Panel on the International Pharmacopoeia and Pharmaceutical Preparations (EAP) inviting comments and posting of the working document on the WHO website for public consultation.	April 2024
Consolidation of comments received and review of feedback. Preparation of working document for discussion.	May – June 2024
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Preparation of a working document for discussion and possible adoption by the ECSPP	August – September 2024
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Any other follow-up action as required.	

WHO good manufacturing practices considerations for the prevention and control of nitrosamine contamination in pharmaceutical products.

47

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- Table of contents
- 49 1. Introduction
- **50** 2. Scope
- **51** 3. Glossary
- 52 4. Points to consider
- 5. Risk assessment
- **54** 6. Root cause analysis
- 55 7. Excipients
- 8. Active Pharmaceutical Ingredients
- 9. Finished Pharmaceutical Products
- 58 10. Acceptable Intake limits
- 59 11. Analytical procedures
- 60 12. Recommendations
- 61 References and further reading

1. Introduction

1.1. Nitrosamines and their precursors are found in food products and other consumer products such as processed meats, alcoholic beverages, and cosmetics. In these cases, they are normally present in small quantities.

- 1.2. Foods such as meats, dairy products and vegetables as well as drinking water may contain low levels of nitrosamines. There is no immediate health risk associated with the use of pharmaceutical products containing levels of a nitrosamine impurity below recommended acceptable intake limits. The actual health risk varies from person to person and also depends on the chemical structure of the nitrosamine contaminant. Nitrosamine impurities may increase the risk of cancer in case of exposure above acceptable levels and over long periods of time (i.e., lifetime intakes below acceptable limit is not expected to significantly increase cancer risks). The risk further depends on several factors, such as:
 - the daily dose of the medication;
 - how long the medication is taken;
 - the level of the nitrosamine impurity in the finished product.

1.3. In recent years, some manufacturers of pharmaceutical products have identified that their products were contaminated with N-nitrosodimethylamine (NDMA), hereafter referred to in general, as nitrosamines). This has led to worldwide recalls of certain products that contained levels of nitrosamines above acceptable limits.

1.4. Nitrosamines is a group or class of compounds which have the chemical structure of a nitroso group bonded to an amine (R1N(-R2)-N=O). The compounds can form by a nitrosating reaction between amines (secondary, tertiary, or quaternary amines) and nitrous acid (coming from nitrite salts under acidic conditions) (1).

1.5. Nitrosating agents include nitrites (e.g. sodium nitrite, NaNO2) and nitrous acid (HNO2), nitric oxide (NO), nitrosyl halides (e.g. CINO, BrNO), dinitrogen trioxide (N2O3), dinitrogen tetroxide (N2O4) and organic nitrites (e.g. t-BuONO). Some can arise from recycled solvents or reused catalysts from different processes or across manufacturing lines with inadequate control and inappropriate monitoring.

97	1.6.	N-Nitrosamines are a class of substances of concern to international regulators and the
98		pharmaceutical industry. This is because many nitrosamines are highly potent mutagenic
99		agents that have been classified as probable human carcinogens. In order to control the
100		presence of nitrosamines in pharmaceutical products, manufacturers should be familiar with
101		the root causes of nitrosamine impurities in their products. A comprehensive risk
102		management plan should be established and implemented.
103		
104	1.7.	Manufacturers should perform risk assessments to determine whether their products are at
105		risk of containing nitrosamine impurities, and ensure that the levels of impurities do not
106		exceed the acceptable limits. Risk assessment should include the assessment of information
107		relating to excipients, active pharmaceutical ingredients (APIs) and finished pharmaceutical
108		product manufacture. It should cover potential formation and presence of nitrosamine
109		impurities, as well as the potential for contamination of other products from e.g. materials,
110		other products or residue on commonly used equipment.
111		
112	1.8.	New impurities of concern may be identified on an ongoing basis. The following nitrosamine
113		impurities are currently of concern: (Note: This not an exhaustive list)
114		N-nitrosodimethylamine (NDMA)
115		 N-nitrosodiethylamine (NDEA)
116		 N-nitrosodiisopropylamine (NDIPA)
117		N-nitroso-N-methyl-4-aminobutanoic acid (NMBA)
118		• 1-methyl-4-nitrosopiperazine (MNP)
119		N-nitrosoethylisopropylamine (NEIPA)
120		N-nitrosodibutylamine (NDBA)
121		
122	1.9.	Materials, equipment and utilities, may contain contaminants that may be carried over into
123		another material, intermediate, excipient or finished product resulting in contamination or in
124		the formation of nitrosamines. This may result in an adulterated product which could be
125		harmful to patients.
126		
127	1.10.	Traces or residue of unwanted substances present in materials, on surfaces of equipment, in
128		the environment, or in carrier material such as water - may be difficult to remove. These may
129		also be difficult to detect through conventional analytical procedures and basic tests.

130		Validated, sensitive, selective analytical procedures may have to be used to detect these
131		contaminants.
132		
133		
134 135	2.	Scope
136	This g	uideline is applicable to all manufacturers of excipients, active pharmaceutical ingredients and
137		ed pharmaceutical products.
138		
139 140	3.	Glossary
141	Accep	table Intake Limit. The maximum Intake level that poses negligible cancer risk, or for serious/life-
142	threat	ening indications where risk and benefit are appropriately balanced. The acceptable intake limits
143	can be	e bound to a specific time-period e.g. daily or cumulative.
144		
145	Carcir	ogenic. Having the potential to cause cancer
146		
147	Maxin	num daily dose. Highest dose per day that has been proven to be safe and effective for the
148	intend	led use, without leading to unacceptable side effects or toxicity.
149		
150 151	Muta	genic. Capable of causing changes or mutations in the genetic material of an organism.
152	Muta	genic impurity. An impurity that has been demonstrated to be mutagenic in an appropriate
153	mutag	genicity test model, e.g., bacterial mutagenicity assay.
154		
155	Nitros	amine. Nitrosamines are organic compounds with the chemical structure $R_2N-N=0$, where R is
156	usuall	y an alkyl group. They feature a nitroso group bonded to a deprotonated amine
157		
158	Nitros	amine impurities. Undesired substances which are formed by the reaction of secondary amines,
159	amide	s, carbamates, derivatives of urea with nitrite or other nitrogenous agents
160		
161	For o	ther definitions, see the WHO Quality Assurance of Medicines Terminology Database - List of
162	Terms	and related guideline (https://www.who.int/publications/m/item/quality-assurance-of-
163	<u>med</u> ic	ines-terminology-database).

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164		
165	4.	Points to consider
166 167	4.1.	Manufacturers of excipients, active pharmaceutical ingredients (APIs) and finished
168		pharmaceutical products (FPPs) should comply with current Good Manufacturing Practices.
169		(2-4).
170		
171	4.2.	Manufacturers should ensure that a pharmaceutical quality system consisting of e.g.
172		procedures, instructions and specifications is in place to ensure the production and control of
173		materials and products that meet safety, quality, purity and efficacy standards.
174		
175	4.3.	Quality risk management should be an important component of the PQS. Manufacturers
176		should identify risks, assess those risks (harm) and implement appropriate controls to
177		eliminate or mitigate those risks.
178		
179	_	
180	5.	Risk assessment
181 182	5.1.	Manufacturers should perform risk assessments to determine whether their products are at
183	5.1.	risk of being contaminated with nitrosamine impurities.
184		Tisk of being contaminated with find osainine imparities.
185	5.2.	The risk assessment should be comprehensive and include but not be limited to the premises,
186		equipment, materials, route of synthesis, production process, interaction between chemicals,
187		excipients, solvents, APIs, packaging components as well as the intended use of the product
188		and route of administration.
189		
190	5.3.	Biological, chemical and physical risks or harms which may be introduced or increased; or
191		should be controlled in each step or stage of production, should be identified.
192		
193	5.4.	An appropriate tool should be used when conducting risk assessment. (5).
194		
195	5.5.	As a minimum, the following basic questions should be considered during the risk
196		assessment:
197		• Is there a possibility of formation of nitrosamine impurities? If so, what are the controls

to reduce/eliminate the formation?

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229

199		How easy is it to detect these nitrosamine impurities?
200		What could be the possible source of the formation of nitrosamine impurities?
201		What is the nature of possible risk(s)?
202		What is the probability of their occurrence?
203		What are the consequences and what is the severity?
204		Is a separate, or dedicated facility or equipment needed?
205		Has an appropriate supplier qualification been done to ensure that there is no risk of
206		contamination of material at the supplier?
207		Are the raw and starting materials, and excipients used, of appropriate purity and
208		quality?
209		
210 211 212 213 214 215	Enviro	naterial Water Excipient API Manufacturing process Nitrosamine impurities nument Equipment Packaging material Stability 1. Ishikawa diagram (Example)*
216	*Includ	e primary, secondary and tertiary causes
217		
218 219	6.	Root cause analysis
220	6.1.	With the identification and assessment of risks for nitrosamine contamination, manufacturers
221		should also do root cause analysis to determine the possible, or probable cause of the
222 223		formation of, or contamination with nitrosamine.
224	6.2.	As a minimum, the following questions should be considered:

Have solvents (fresh and recovered) been considered for possible contamination?

did you perform an on-site assessment on contracted facility? Did you assess potential

What is the quality and purity of the solvent used in any step of the processing?

risk of contamination and cross-contamination during the recovery of solvents?

Are solvents recovered (on-site or off-site/contracted out)?

230	• Is there an appropriate procedure in place to ensure purity of the solvent obtained from
231	the recovery process?
232	 Is any nitrate such as sodium nitrate used, including in reagents and catalysts?
233	Is any nitrosating agent used?
234	• Is there any risk that nitrates/nitrosating agents can be generated as an impurity during
235	the manufacturing process?
236	 Are there test results for materials showing nitrites, nitrates and nitrosamines?
237	Has water been tested for the presence of potential nitrosamine forming agents, such
238	as chloramines, nitrites and nitrates?
239	• Is there any secondary or tertiary amine present in the manufacturing process, e.g. raw
240	materials, intermediate, reagent, solvent?
241	• Is there any amide, amine or ammonium salt present in the substance(s) e.g. raw
242	materials, intermediate, reagent, solvent?
243	• Have utilities such as water been considered as a possible source of contamination?
244	Have equipment been considered as a possible source of contamination, including
245	efficiency of cleaning procedures?
246	• Are nitrites (NO2-), nitrous acid, nitrates (NO3-), nitric acid, or azides (N3-) or their
247	sources present in any excipients (e.g., microcrystalline cellulose), processing aids (e.g.,
248	water, nitrogen)?
249	 Are peroxides present in any of the excipients, processing aids?
250	• Are nitrites (NO2-), nitrous acid, nitrates (NO3-), nitric acid, or azides (N3-) or their
251	sources present in packaging components (including ink, and materials permeability
252	factors)?
253	• Is there a risk that secondary or tertiary amine-contaminants may be present in any
254	primary amines used in your manufacturing process?
255	 Are any components containing/potentially containing nitrites and amines present
256	together in solution or in suspension during processing (e.g., during granulation,
257	coating)?
258	• Are nitrites (NO2-), nitrous acid, nitrates (NO3-), nitric acid, or azides (N3-) or their
259	sources present in chemically synthesized APIs?
260	Based on the structure of drug substance and excipients, is there any possibility of
261	formation of nitroso compounds by interaction of drug substance and excipients?
262	Are any components containing/potentially containing nitrites and amines maintained

together at elevated temperatures (e.g., during drying, coating stages, autoclaving)?

264	 Do solvents or any other process materials undergo recycling/recovery?
265	 In the manufacturing process of the drug product, are any of the solvents, spent
266	solvents, or process materials treated prior to or during recovery (in-house or by a third
267	party) such that the treatment could lead to formation of amines or nitrosonium ions
268	that could be introduced back into the process through the recovered solvents?
269	 Are the recovered materials, if any, dedicated to the process?
270	 Is there a potential for nitrosamine impurity formation during the finished product
271	manufacturing, through degradation and by-products (i.e., if certain excipients, APIs, or
272	packaging components containing sources of amines and nitrite are used together)?
273	 Are "sartan" products manufactured in the same facility? Is there a risk of cross-
274	contamination?
275	 Are manufacturing equipment material of construction of any concern?
276	 Are chemicals such as sodium azide or sodium nitrite, which are primary sources of
277	nitrosamine impurity, used in the facility?
278	• Are cleaning procedures of equipment involved in manufacturing validated using worst-
279	case product consideration (i.e., solubility, potency, toxicity and cleanability)?
280	 Are there amines and nitrosonium ions (degradation and by-products) likely to come
281	into contact with each other either in the same processing step or through carryover
282	into subsequent processing steps?
283	 Is there any potential of nitrosamine formation during storage throughout the finished
284	product's shelf life?
285	 Is chloramine used as part of the water treatment process, for water used for cleaning,
286	or as part of the production process?
287	 Have the cleaning solvents/cleaning agents used, been assessed for nitrosamine or
288	nitrosamine precursor risk?
289	 Does any of the manufacturing processes contribute toward formation of N-
290	Nitrosamines?
291	 Is there any risk of nitrosamine formation due to the use of nitrogen?
292	
293	6.3. Examples of possible root causes are listed below. Appropriate controls should be identified
294	and implemented to mitigate risks:
295	
296	6.3.1. Amines and nitrite reaction

Formation of nitrosamines is possible in the presence of secondary, tertiary, or quaternary amines and nitrite salts under acidic reaction conditions. Under these conditions, nitrite salts may form nitrous acid, which can react with an amine to form a nitrosamine. Nitrites used as reagents in one step can carry over into subsequent steps, despite purification operations, and react with amines to generate nitrosamine impurities. Therefore, whenever nitrite salts are present, carryover into subsequent steps cannot be ruled out. In general, processes that use nitrites in the presence of secondary, tertiary, or quaternary amines are at risk of generating nitrosamine impurities (1).

6.3.2. Amine functional groups in processing

Amines are sometimes added as reagents or catalysts during a manufacturing process. Nitrosamine formation is possible when amines react with nitrous acid or other nitrosating agents. Another source of secondary amines is amide solvents. These are susceptible to degradation under certain reaction conditions. (Note the degradation of N,N-dimethylformamide, N-methylpyrrolidone, N,N-dimethylacetamide, and N,N-diethylacetamide).

6.3.3. Introduction of Nitrosamine impurities

Nitrosamine impurities can be introduced into materials and products when contaminated materials such as starting materials and raw materials, are incorporated into products. Starting materials and intermediates may be at risk through cross-contamination if they are manufactured at sites where nitrosamine impurities are formed during other processes. In addition, materials are sometimes contaminated during storage, shipment, distribution.

6.3.4. Solvents

Fresh solvents can be contaminated at different stages in the supply chain, as well as during transfer between storage vessels. Recovered materials such as solvents, reagents, and catalysts may also pose a risk of nitrosamine impurities due to the presence of residual amines (such as trimethylamine or diisopropylethylamine). The use of recovered

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331 solvents that are comingled from different processes or across manufacturing lines 332 without control and monitoring can introduce nitrosamine impurities. Outsourcing of 333 recovery of raw materials (e.g., solvents, reagents, and catalysts) can pose a risk of 334 contamination. 335 336 337 6.3.5. Inadequate equipment cleaning 338 The suitability of use of equipment (and management of utilities), as well as their 339 340 cleaning, should be assessed to ensure that the risk of introducing nitrosamine impurities 341 through their use, is appropriately controlled. 342 Materials, intermediates and products can be contaminated if adequate cleaning of 343 equipment between different materials, products or batches is not carried out, or is not 344 345 validated as being capable of removing residue or impurities of concern. 346 347 Inadequate and unvalidated cleaning procedures can also lead to cross-contamination if precautions to avoid nitrosamine contamination are not in place before materials are 348 349 combined for recovery. 350 351 The nature and composition of the cleaning solvents used for the cleaning of reactors 352 used during API synthesis/purification and for the cleaning of finished dosage form 353 equipment should be considered as cleaning solvents (e.g. amines) could react to form nitrosamines under certain conditions if the equipment is not perfectly dry prior to its 354 355 used for subsequent manufacture and/or if there are residues remaining. The 356 contaminants found in the cleaning solvents, could also react to form nitrosamines (6). 357 6.3.6. Utilities 358 Utilities such as HVAC and water systems, may also be a source of contamination. Risk 359 360 assessment should be done to consider the contaminants in air and water, as well as the

treatment of air and water - as these contain nitrites and other contaminants.

363		Potable water is sometimes used in the production of materials such as excipients and APIs;
364		or to clean equipment. Water may contain low levels of chloramine and or nitrites/nitrates,
365		which are known to potentially react with secondary amines to form nitrosamine
366		impurities, depending on specific conditions. The source, quality and purification of water
367		may impact on the absence, presence or formation of nitrosamine impurities. For example,
368		chlorination may contribute to the formation of nitrosamine impurities. Chloramine,
369		nitrite/nitrate and nitrosamine levels in water should thus be determined.
370		
371		Where required, water should be purified to remove unacceptable impurities before use.
372		
373	6.3.7.	Nitrosamines from environmental contamination.
374		
375		Atmospheric NO2 is a nitrosating agent for various secondary amines, such as DMA. It has
376		been shown that there is a correlation between the concentration of atmospheric NO2 and
377		the NDMA content in certain products. The following examples are controls that could be
378		considered:
379		• Inlet air to equipment, such as fluidized bed friers, should be appropriately
380		controlled as it may be contaminated;
381		 APIs with low DMA content should be used where possible;
382		Risks associated with process parameters such as granulation drying time and
383		temperatures should be controlled.
384		
385	6.3.8.	Quenching Process as a Source of Nitrosamine Contamination
386	0.5.0.	Quenering Process as a source of Microsamine Contamination
387		The risk of nitrosamine formation when a quenching step is performed directly in the main
388		reaction mixture should be avoided or controlled.
389		reaction mixture should be avoided or controlled.
390		Inadequate removal of impurities, or operations which are not optimized for removing
391		specific impurities of concern, may increase the risk of nitrosamine impurities carried over
392		to subsequent steps.
393		to subsequent steps.
394	6.3.9.	Poorly controlled reaction conditions
395		

396 The manufacturing process for APIs should be optimized. Reaction conditions such as 397 temperature, pH, or the sequence of adding reagents including catalysts, intermediates, or 398 solvents should be appropriate and controlled to prevent the formation of impurities. 399 **Excipients and packaging material** 7. 400 401 402 7.1. Excipients should be manufactured in compliance with WHO GMP for excipients used in 403 pharmaceutical products (2). 404 405 7.2. Impurities, such as nitrite/nitrate, can be found in a range of commonly used excipients. This 406 may lead to nitrosamine impurities forming in pharmaceutical products during production and 407 storage of the product. The supplier qualification program should cover the verification of 408 controls over the possibility of nitrite impurities. 409 7.3. 410 Packaging materials may be a source of contamination. Nitrocellulose in PTP aluminium 411 printing ink is commonly known as a nitrosating agent. 412 7.4. Where the excipient is identified as a probable cause for formation of nitrosamine impurities, 413 appropriate controls should be implemented. This may include consideration to change the 414 415 supplier of the excipient or the change of the excipient to reduce the risk of nitrosamine impurities formation. 416 417 Note: See reference to water, under the section "utilities" 418 419 **Active Pharmaceutical Ingredients (APIs)** 8. 420 421 8.1. APIs should be manufactured in compliance with WHO GMP for APIs. 422 423 424 8.2. API manufacturers should carefully design route of synthesis (ROS) to minimize or prevent the 425 formation of nitrosamine impurities. (3, 7-9). 426 427 8.3. Reaction conditions that may produce nitrosamines should be avoided as far as possible, 428 starting from the process development stage. Where this is not possible, the process should

429		be adequately controlled and should be capable of consistently reducing nitrosamine
430		impurities.
431		
432	8.4.	Bases other than secondary, tertiary, or quaternary amines (when possible) should be used if
433		ROS conditions may form nitrosamines.
434		
435	8.5.	Caution should be used when the ROS involves the use of amide solvents (e.g., N,N-
436		dimethylformamide, N,N-dimethylacetamide, and N-methylpyrrolidone).
437		
438	8.6.	Where possible, nitrites should be replaced with other quenching agents for azide
439		decomposition processes.
440		
441	8.7.	Sequences of reactions, processes, and reaction conditions (such as pH, temperature, and
442		reaction time) should be optimized and consistently controlled for avoiding the formation of
443		nitrosamine impurities.
444		
445	8.8.	Manufacturing process should be designed to facilitate the purge of nitrosamine impurities in
446		the subsequent processing steps
447		
448	8.9.	Supply chains should be audited and monitored for any at-risk raw materials, starting
449		materials, and intermediates.
450		
451	8.10.	Records including the name of the raw material manufacturer and its supplier, roles of the
452		actual manufacturers of such materials, and any re-packers and distributors who handle the
453		materials before API manufacture, should be maintained.
454		
455	8.11.	When appropriate, controls and additional specifications should be considered for at-risk
456		materials to prevent nitrosamine contamination.
457		
458	8.12.	API manufacturers should verify with their suppliers whether the purchased materials used in
459		their processes are recovered.
460		

461	8.13.	Recovered materials such as solvents, reagents, and catalysts should be used only in the same
462		step or in an earlier step (if there is sufficient purification) of the same process from which it
463		was collected.
464		
465	8.14.	The recovered materials should meet appropriate standards before reuse. If the recovery of
466		materials is outsourced to third-party contractors, the API manufacturer should audit the
467		contractors' validation of procedures, including cleaning procedures.
468		
469	8.15.	Potable water may contain low levels of nitrite and even nitrosamines from environmental
470		contamination. Nitrite and nitrosamine levels in water should be determined. Where
471		required, water should be purified to remove unacceptable impurities before use.
472		
473	8.16.	API batches containing nitrosamine impurities may be reprocessed or reworked under
474		oversight of the quality unit. Records should be kept.
475		
476	8.17.	Batches of API containing levels of nitrosamine impurities above the recommended limits
477		should not be released for sale or distribution.
478		
479	8.18.	Batches of API with unacceptable levels of nitrosamine impurities already in distribution,
480		should be reported to the national medicine regulatory authority. A batch or product recall
481		should be considered.
482		
483	9.	Finished Pharmaceutical Product (FPP)
484		manufacturers
485		
486	9.1.	Products should be manufactured in compliance with WHO GMP for pharmaceutical products
487		(4).
488		
489	9.2.	Risk assessments should be conducted to determine the potential for nitrosamine impurities
490		in FPPs.
491 492	9.3.	A control strategy should be defined to prevent or mitigate the risk of nitrosamine
493	2.2.	contamination of FPPs.
494		

495	9.4.	The risk assessment should include evaluation of the supply chain, any excipient, API
496		processing, utilities, as well as storage, re-packaging, distribution pathway and degradation
497		that may introduce nitrosamines during production or storage. Consideration should be given
498 499		to establish whether nitrosamines could form in an FPP, over the product's shelf life.
500	9.5.	If a risk of nitrosamine presence is identified, confirmatory testing of batches should be
501		conducted using sensitive, appropriately validated, analytical methods.
502		
503	9.6.	If a nitrosamine impurity is detected, the root cause should be determined. Where
504		appropriate, changes in the manufacturing process to mitigate or reduce the nitrosamine
505		impurities should be made.
506		
507	9.7.	The risk of nitrosamine impurity formation during the manufacture and packaging of the
508		finished pharmaceutical product (such as when certain containers, API or packaging
509		components come into contact with amines or nitrites, e.g. reaction of secondary amines in
510		printing inks with certain nitrocellulose lacquers or coating materials when heated) should be
511		considered in the risk assessment.
512 513 514 515 516 517	9.8.	Processing steps such as granulation or drying may increase the risk of nitrosamine impurity formation. Where appropriate, changes in the manufacturing process to mitigate or reduce the nitrosamine impurities should be made.
517 518 519 520 521 522		Note: Purification steps during the production of an API may assist in mitigating risks of the presence of nitrosamine impurity in the API. This may not be the case with the production steps of a finished pharmaceutical product.
523	10.	Acceptable Intake (AI) limits
524 525	10.1.	The low levels at which the nitrosamine impurities occur create challenges for testing.
526		
527	10.2.	Appropriate procedures should be developed and validated. (See also methods
528		recommended by SRAs). Note: Higher temperature conditions of some test methods may
529		cause the sample to generate NDMA.
530		

531	10.3.	Generally, sensitive methods with limits of quantitation (LOQ) in the parts-per-billion (ppb)
532		should be used. The LOQ and limit of detection (LOD) should be as low as reasonably practical
533		for products for which the maximum daily dose is high (e.g., greater than 1 g).
534		
535	10.4.	Where more than one nitrosamine listed in appendix 1 is detected, the analytical procedure
536		should be validated for LOQs below 0.03 ppm to accurately quantify a total nitrosamine level
537		of not more than 26.5 ng/day. (For example, if the MDD is 1200 mg, the LOQ should be below
538		0.02 ppm. FDA's public webpage includes validated analytical test methods recommended for
539		detecting nitrosamine impurities in several different APIs and products) (1).
540		
541	10.5.	Only limited impurity-specific toxicity data is available for NDMA and NDEA. Based on this
542		information interim acceptable intakes for these specific impurities have been adopted by
543		most major regulators.
544		Acceptable Intake (AI) limits for nitrosamines in FPPs (10)
545	10.6.	Al limits should be established. (Note: Different approaches are described in the literature
546		and guidelines as those published by ICH, Health Canada, and the US FDA). For example:
547		
548	10.	6.1. If N-nitrosamines are identified with sufficient substance specific animal carcinogenicity
549		data, the TD50 should be calculated and used to derive a substance specific limit for
550		lifetime exposure as recommended in ICH M7 guideline (9);
551		
552	10.	6.2. If N-nitrosamines are identified without sufficient substance specific data to derive a
553		substance specific limit for lifetime exposure as described above, the Carcinogenic
554		Potency Categorization Approach (CPCA) for N-nitrosamines should be used to establish
555		the AI, unless other robust data are available that would override this AI;
556		
557	10.	6.3. A negative result in an GLP-compliant Enhanced Ames Test (EAT) allows control of the N-
558		nitrosamine at 1.5 $\mu g/day$. For substances testing positive, the AI should be established
559		using options 1 or 3;
560		

561	10.	6.4. If a surrogate nitrosamine is available with sufficiently robust carcinogenicity data, the
562		TD50 from the surrogate substance can serve as a point of departure for derivation of Al
563		by SAR and read across.
564		
565	10.7.	A negative result in a relevant well-conducted in vivo mutagenicity study can allow control of
566		the N-nitrosamine as a non-mutagenic impurity, i.e., according to Q3A/B limits, irrespective of
567		the limit calculated through option 1, 2 or 3. For substances testing positive, the AI should be
568		established using options 1 or 3.
569		
570	10.8.	In setting AI limits for nitrosamines, consideration should be given to:
571		• the Enhanced Ames Test (EAT) conditions and the Carcinogenic Potency Categorization
572		Approach (CPCA);
573		• the threshold below which a nitrosamine impurity is not expected to be included in routine
574		testing specifications;
575		• testing approaches where more than one strength of a dosage form is concerned, and
576		• expectations when a nitrosamine impurity cannot be synthesized or isolated and purified.
577		
578	10.9.	Recommended AI limits are presented in literature. (The AI limit is a daily exposure to a
579		compound such as NDMA, NDEA, NMBA, NMPA, NIPEA, or NDIPA that approximates a
580		1:100,000 cancer risk after 70 years of exposure. See ICH M7 (9) and USA FDA regulatory
581		information note ¹)
582		
583	10.10.	Examples of Interim allowable daily intake limits for a selection of N-nitrosamine impurities
584		are presented in Annex 1. For a nitrosamine impurity that is not included in the appendix, the
585		principles as outlined in ICH's M7 guideline are recommended to be used to determine an
586		acceptable Intake (9).
587		
588	10.11.	The conversion of AI limit into ppm varies by product and is calculated based on a product's
589		maximum daily dose (MDD) as reflected in the drug label (ppm = AI (ng)/MDD (mg)). These
590		limits are applicable only if a drug product contains a single nitrosamine (1).
591		
592		

 $^{^{1}\ \}underline{\text{https://www.fda.gov/regulatory-information/search-fda-guidance-documents/updated-information-recommended-acceptable-intake-limits-nitrosamine-drug-substance-related}$

11. Analytical procedures

11.1. Validated analytical procedures should be used when testing for the presence of
 nitrosamines. The procedure should be sensitive for the determination of the specific
 nitrosamine(s) in the product.

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11.2. Where the presence of a nitrosamine is confirmed, it should not exceed the acceptable limits. Where it exceeds the acceptable limit, appropriate corrective action should be taken. If the manufacturing procedure needs to be changed, the change management procedure should be followed for the relevant variation. In this case, where relevant, the NMRA should be informed.

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Where appropriate, analytical methodology to separate thermally labile nitrosamine
 impurities using gas chromatography (GC) coupled with detection by thermal energy analysis
 (TEA) or mass spectrometry (MS), should be used.

608

Liquid chromatography (LC) coupled with detection by TEA, MS, or ultraviolet light (UV) may
 be an alternative analytical methodology applicable to both volatile and non-volatile
 nitrosamines.

612

High performance liquid chromatography (HPLC) with UV detection has low sensitivity and
 may only be adequate for analysis of low dose drugs with lower limits.

615

616 11.6. Examples of analytical procedures are presented in table 1.

617

Table 1. Examples of analytical procedures*

Methodology	Detector	Limit of Detection
Gas chromatography	TEA	< 0.1 to 5 ppb
	MS	1 to 5 ppb
Liquid chromatography	TEA	1 to 50 ppb
	MS	1 to 5 ppb
	UV	1 to 200 ppb

^{*} used for detection of nitrosamine in e.g. water and food products

	rage 21
12	Recommendations
12.1. 12.1.	Excipient, API and FPP manufacturers should take steps to mitigate the risk of nitrosamine impurities in their products.
i 12.2.	Risk assessment of nitrosamine impurities should be conducted in a timely manner, as early as during product development as well as thereafter during the manufacturing of excipients, APIs and FPPs.
12.3.	 average nitrosamines and potential nitrosamine precursors content and batch to batch variance differ among excipients; for solid dosage forms, the nitrosamine and potential nitrosamine precursors contribution is dominated by the highest formula % excipients, e.g., the fillers (diluents), which are typically used in larger proportion, and are characterized by low nitrite levels and low variability, leading to an average value of 1 µg/g nitrite in a typical formulation; substantial differences may occur in average nitrosamine and potential nitrosamine precursor content in batches from different excipient vendors potentially reflecting differences in source materials or processing methods for excipient manufacturing; selection of raw materials or processing by excipient manufacturers may help reduce nitrite levels in finished drug product formulations, and thus the overall risk of nitrosamine formation in cases where the product contains vulnerable amines (11).
12.4.	The benefit and the risks of products with levels of nitrosamines exceeding acceptable limits or more than one nitrosamine should be reviewed by the NMRA. When considering the withdrawal, the NMRA should balance the impact on the patient if the product will no longer be available. This should involve determining the availability of alternative products or treatments on their own market and the clinical impact of stopping or switching to a different treatment.

Where manufacturers identify contamination of nitrosamines above acceptable limits,

appropriate action should be taken. Risk and impact assessment should be done with root

cause determination. Thorough investigations should be done to identify whether shared

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12.5.

654	facilities, shared equipment or other batches may be impacted, whether common excipients,
655	starting materials or solvents were used which may be the potential source of contamination
656	The investigation should be extended to other batches which may have been impacted.
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Appendix 1: Examples of Interim allowable daily intake limits for a selection of N-nitrosamine impurities.

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Impurity	Chemical name	Allowable daily intake
(Abbreviation)		
NDMA	N-nitrosodimethylamine	96.0 ng/day
NDEA	N-nitrosodiethylamine	26.5 ng/day
NMBA	N-nitroso-N-methyl-4-aminobutryric-acid	96.0 ng/day
DIPNA	N-nitroso-diisopropylamine	26.5 ng/day
EIPNA	N-nitroso-ethylisopropylamine	26.5 ng/day

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